Nucleosides and Nucleotides. 114. A Convenient Method for the Synthesis of 3-Deazapurine Nucleosides from AICA-Riboside¹

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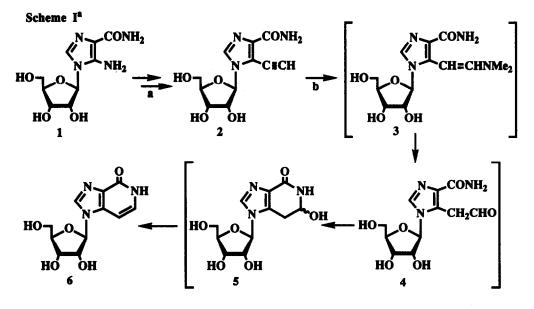
Key Words: nucleoside; 3-deazapurine nucleoside; 3-deazaadenosine; 3-deazaguanosine; 3-deazainosine; acetylene; antiviral agent

Abstract: 3-Deazapurine nucleosides, 3-deazainosine (δ), 3-deazaguanosine (12), and 3-deazaadenosine (17), were synthesized from 5-ethynyl-1- β -D-ribofuranosylimidazole-4-carboxamide (2) or -4-carbonitrile (13), which were readily obtained from AICA-riboside, via intramolecular ring closure.

Since the original isolation of the antibiotics toyocamycin and tubercidine, and subsequent structural elucidation of these antibiotics as 7-deazapurine nucleosides, numerous deazapurine nucleosides have been synthesized in the search for anticancer and antiviral agents. Interesting compounds in this category are the 3deazapurine nucleosides, especially 3-deaza-adenosine and -guanosine. 3-Deazaadenosine was found to be a potent inhibitor of S-adenosylhomocysteine hydrolase.² It was also reported that the nucleoside inhibits the reverse transcriptase of HIV-1.3 On the other hand, 3-deazaguanosine has demonstrated potent antiviral activity in vitro against a variety of DNA and RNA viruses⁴ as well as in vivo activity against L1210 leukemia and adenocarcinoma 755 in mice.⁵ 2'-Deoxy analogues of 3-deazapurine nucleoside were also used as valuable probes for study of protein-nucleic acid interactions.⁶ In spite of the interesting biological and pharmacological properties of the 3-deazapurine nucleosides, almost all of the synthetic methods so far reported involved classical condensation methods with appropriate 3-deazapurines and sugars, which suffer from regio- and stereochemical disadvantages.⁷ Therefore, it seems worthwhile to develop a new general synthetic route from readily available nucleosides which will be free from these disadvantages. To achieve our purpose, a most straightforward route is to synthesize 4-carboxamidoimidazole nucleosides having formyland cyano-methylene groups at the 5-position, followed by ring closure of these nucleosides to make the target nucleosides. We have already reported the introduction of alkynyl groups into the 5-position of such imidazole nucleosides, which was easily obtained from 5-amino-1-B-D-ribofuranosylimidazole-4-carboxamide (AICAriboside, 1), by organopalladium chemistry,⁸ Since modification of terminal alkynes has already been demonstrated as a Willgerodt-Kindler reaction,⁹ we apply this methodology to 5-ethynyl-1- β -Dribofuranosylimidazole-4-carboxamede (2) and -4-carbonitrile (13) to find out a new convenient method for the synthesis of 3-deazapurine nucleosides.

Treatment of 2 with 50% aqueous dimethylamine in EtOH at 80 °C in a sealed tube, followed by 50% aqueous acetic acid—EtOH (1: 1) gave the desired 3-deazainosine (6)¹⁰ in 64% yield. In this reaction, a series of reactions (hydroamination of 2, hydrolysis of 3, intramolecular ring closure of 4, and dehydration from 5) could proceed successively. Treatment of 5 with acetic acid accelerated the formation of 6.



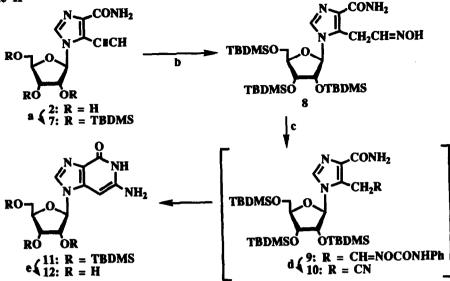


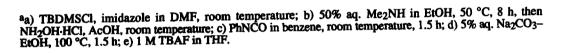
a) ref. 8; b) 50% aq. Me₂NH in EtOH, 80 °C, 5.5 h, then 50% aq. AcOH in EtOH, room temperature.

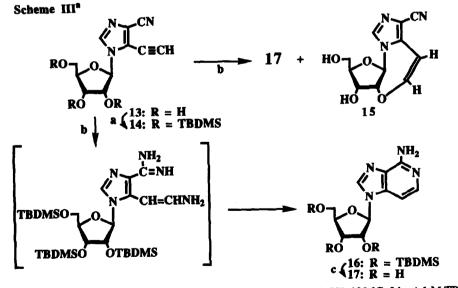
For the synthesis of 3-deazaguanosine (12), it could be easily prepared if the formyl group in 4 reacts with hydroxylamine faster than the intramolecular ring closure to 5. Compound 2 was first silylated with *tert*butyldimethylsilyl chloride (TBDMSCl) in the presence of imidazole in N, N-dimethylformamide (DMF) to give 7 in 92% yield. After hydroamination of 7 was carried out at 50 °C with 50% aqueous dimethylamine, hydroxylamine hydrochloride and acetic acid were then added to the reaction mixture. The desired 8 was obtained in 91% yield without formation of the 3-deazainosine derivative. After several attempts, we found phenyl isocyanate as the choice for dehydration. Treatment of 8 with phenyl isocyanate in benzene gave intermediate 9, which was subsequently heated in EtOH-5% aqueous sodium carbonate at 100 °C to give 11 in 56% yield from 8 via intermediate 10. Compound 11 was then deprotected by tetrabutylammonium fluoride (TBAF) to give 3-deazaguanosine (12)¹⁰ in 78% yield.

The route for the synthesis of 3-deazaadenosine (17) was similarly based on the knowledge that intramolecular ring closure proceeds between substituents of the imidazole ring. That is to say, it could be prepared if both hydroamination to 5-ethynyl group and conversion of the 4-cyano group to the corresponding 4-amidino group proceed at the same time. We first used 13 as a starting material, which was obtained in a similar manner as 2^{8b} , with methanolic ammonia at 120 °C in a sealed tube. The desired 17 was obtained in only 17% yield along with 5, 2'-O-cycloetheno derivative 15^{11} , which might be obtained by a direct reaction between the 2'-hydroxyl and the 5-ethynyl groups. In this reaction, it is thought that the transformation of the cyano group to the amidino group would be a rate-determining step, therefore, formation of 15 is of greater advantage than of 17 in unprotected form. Consequently, the hydroxyl groups of the sugar moiety were protected with TBDMSCl to give 14 in 83% yield. When 14 was treated with methanolic ammonia as above, 3-deazaadenosine derivative 16 was obtained in 76% yield. Compound 16 was then deprotected to give 3-deazaadenosine (17).¹⁰ In this method, 17 was obtained as a free form which is free from the perplexing problems reported by Mizuno *et al.* before.¹²









a) TBDMSCl, imidazole in DMF, room temperature; b) NH3 / MeOH, 120 °C, 3 h; c) 1 M TBAF in THF.

We concluded that 3-deazapurine nucleosides, 3-deazainosine, 3-deazaguanosine, and 3-deazaadenosine, were conveniently synthesized from AICA-riboside. This method can be generalized and used for the synthesis of a wide variety of 3-deazapurine derivatives.

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References and Notes

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- 10. Physical constants of 6, 12, and 17 were identical in all respects with those reported.⁷
- 15: mp 198-201° C; MS m/z 249 (M⁺); ¹H NMR (DMSO-d₆) δ 7.77 (s, 1H, H-2), 6.79 (d, 1H, J = 7.7 Hz, vinyl proton), 5.84 (d, 1H, J = 4.4 Hz, 3'-OH), 5.69 (d, 1H, J = 6.0 Hz, H-1'), 5.66 (d, 1H, vinyl proton), 4.96 (dd, 1H, J = 5.0, 6.0 Hz, 5'-OH), 4.40 (dd, 1H, J = 4.9 Hz, H-2'), 4.34 (m, 1H, H-3'), 4.24 (m, 1H, H-4'), 3.48 (m, 2H, H-5', 5"); Anal. Calcd for C₁₁H₁₁N₃O₄: C, 53.01; H, 4.45; N, 16.86. Found: C, 52.99; H, 4.48; N, 17.01.
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